D 120544	(Pages: 2)	Name
		Reg. No

FOURTH SEMESTER (CBCSS—UG) DEGREE EXAMINATION APRIL 2025

Chemistry

CHE4C04—PHYSICAL AND APPLIED CHEMISTRY

(2019—2023 Admissions)

Time : Two Hours

Maximum : 60 Marks

Section A (Short Answers)

Answer questions up to 20 marks.

Each question carries 2 marks.

- 1. What is an emulsion? Give an example.
- 2. Explain electrophoresis.
- 3. Write any *four* advantages of paper chromatography.
- 4. Define Rf value? How can it be employed in the identification of a compound?
- 5. Give two examples each for food preservatives and artificial sweeteners.
- 6. Write the composition of LPG and CNG.
- 7. Write any two uses of Kevlar and Nomex.
- 8. Which are the monomers of Buna-N and Nylon 66.
- 9. Explain bio-magnification and bioaccumulation.
- 10. What is air pollution? Mention any *two* air pollutants.
- 11. What is meant by chemical shift?
- 12. Explain red shift and blue shift in UV-Visible spectroscopy.

(Ceiling of marks: 20)

Turn over

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Section B (Paragraph)

Answer questions up to 30 marks.

Each question carries 5 marks.

- 13. Describe the principle and applications of thin layer chromatography.
- 14. Write a note on the various applications of nanomaterials.
- 15. State and explain Hardy Schulze rule.
- 16. What are the different types of electronic transitions in molecules? Arrange them in the order of increasing energy.
- 17. Depict the structure and write any two applications each of Teflon and Bakelite.
- 18. What is acid rain? What are the causes and effects of acid rain.
- 19. What is a dye? Draw the structure and mention any *two* applications of alizarin.

(Ceiling of marks: 30)

Section C (Essay)

The any one question.

Each question carries 10 marks.

- 20. (i) Describe briefly the principle involved in NMR spectroscopy.
 - (ii) Explain spin-spin coupling taking an example.
- 21. Write a note on the sources, effects and control measures of thermal pollution.

 $(1 \times 10 = 10 \text{ marks})$