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		Reg. No

FIRST SEMESTER (CUFYUGP) DEGREE EXAMINATION NOVEMBER 2024

Chemistry

CHE1CJ101—INORGANIC CHEMISTRY—I

(2024 Admission onwards)

Time: Two Hours

Maximum: 70 Marks

Section A

All question can be answered.

Each question carries 3 marks.

(Ceiling: 24 marks).

- 1. Write Born-Lande equation and explain the terms used.
- 2. "Dipole moment of ${\rm CCl_4}$ is zero" why?
- 3. What are the factors affecting the formation of ionic compounds?
- 4. Write some examples of nanomaterials in water purification with their mechanism of action.
- 5. What are the different types of nanomaterials used in solar cells?
- 6. How do surface area to volume ratio of nano materials influence their properties?
- 7. What are the key steps in providing first aid for electric shock?
- 8. How would you use a fire extinguisher to put out fire in the laboratory?
- 9. What are the health effects of inhaling poisonous gases?
- 10. Distinguish between molarity and molality.

Turn over

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Section B

2

All question can be answered.

Each question carries 6 marks.

(Ceiling: 36 marks).

- 11. Explain three fundamental points of Fajans rule with examples.
- 12. "Oxygen molecule is paramagnetic but not nitrogen molecule". Explain with MO diagram.
- 13. How many lone pairs and bond pairs are present in ${\rm XeO_3}$? Justify your answer with hybridization.
- 14. How can we minimize measurement errors?
- 15. Explain different types of errors.
- 16. What are the methods of representing precision?
- 17. How are nanomaterials used in the removal of dyes from waste water?
- 18. What are novel properties of nanomaterials that are not seen in bulk materials?

Section C

Answer any **one** questions. Each question carries 10 marks. $(1 \times 10 = 10 \text{ marks})$

- 19. You are working with a new chemical in the lab. How do you use MSDS to ensure safe handling?
- 20. Discuss the hybridization and shape of ClF₃, BrF₅, and SF₄.